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ORGANIC PHOSPHORUS COMPOUNDS 92.1 SYNTHESIS AND PROPERTIES OF AZETIDINE-3-PHOSPHONIC- AND 3-PHOSPHONOUS ACID

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Azetidine-3-phosphonic acid, 4a, and azetidine-3-phosphonous acid, 4b, have been prepared by the interaction of 1-benzhydryl-3-methanesulfonato-azetidine with sodium phosphite or sodium O-ethyl-diethoxymethylphosphonite, followed by catalytic debenzhydrylation with H_2 and hydrolysis. Neither acid shows biological activity.

Key words: 3-Diethylphosphonylazetidine; 3-(O-ethyl-diethoxy-methylphosphinyl)azetidine; azetidine-3-phosphonic acid; azetidine-3-phosphonous acid; debenzhydrylation.

INTRODUCTION

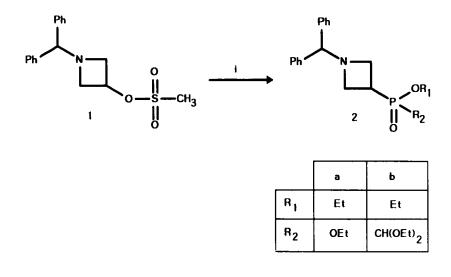
L-Azetidine-2-carboxylic acid occurs in nature² and has been found to inhibit the growth of E. coli cultures and various seedlings³ and to cause abnormalities in growing embryos.⁴ Azetidine-3-carboxylic acid is also known⁵ and has been shown to have gameticidal properties, i.e., it sterilizes male anthers in plants.⁶ Since replacement of a carboxylic acid function in biologically important molecules by a phosphorus acid often provides biologically active molecules (see e.g., References 7 and 8) it seemed of interest to synthesize azetidine-3-phosphonic and 3-phosphonous acid and determine their biological activity.

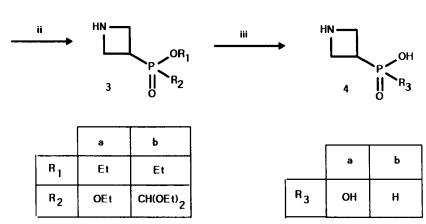
RESULT AND DISCUSSION

Interaction of 1-Benzhydryl-3-methanesulfonato-azetidine (1)⁵ with sodium phosphite or sodium O-ethyl-diethoxy-methylphosphonite in dimethylformamide gave 1-benzhydryl-3-O,O-diethylphosphonylazetidine, 2a, and 1-benzhydryl-3-(O-ethyldiethoxymethylphosphinyl)azetidine 2b in 31.6 and 40% yield, respectively. Both products could be purified by chromatography on silica-gel (Scheme). Debenzhydrylation of 2a and 2b with H_2 over a 5% Pd/C catalyst in ethanol proceeded at normal pressure and ambient temperature and gave 3a or 3b in about 60% yield.

Dealkylation of 3a was successfully achieved with trimethylsilylbromide in chloroform solution at room temperature. Treatment of the silylester with isopropanol/propylene oxide gave azetidine-3-phosphonic acid 4a as white crystals in 82.7% yield. The ³¹P-chemical shift of 18.18 ppm proves its structure as a phosphonic acid.

The corresponding azetidine-3-phosphonous acid 4b was obtained also as white





i: NaH/HP(O)OR1R2/DMF

ii: Pd/C,H2,EtOH/HCl

iii: (CH3)3SiBr/CHCl3/Propylene oxide/Isopropanol or HCl 2N / Propylene oxide/Isopropanol

Scheme

crystals in 65.8% yield by hydrolysis of 3c with 2N HCl at reflux temperature and purification with propylene oxide/methanol.

The ³¹P-chemical shift of 24.3 ppm with a P—H coupling constant of 525.9 Hz proves it structure as a phosphonous acid.

Biological Activity

In contrast to azetidine-3-carboxylic acid⁶ the phosphonic, **4a**, and phosphonous acid analog **4b**, show no antifungal, herbicidal or gameticidal activity.

EXPERIMENTAL

Phosphorus NMR-spectra were recorded using a Bruker WP80 spectrometer at 32.28 MHz (Reference 85% H₃PO₄) and 'H-NMR spectra were recorded with a Varian EM 360 spectrometer at 60 MHz or a Bruker WM 250/250 MHz spectrometer (Reference (CH₃)₄Si). The chemical shifts are reported in ppm, with negative values being upfield of the standard, and positive downfield.

1. 1-Benzhydryl-3-diethylphosphonylazetidine, 2a. To 70.1 g (0.508 mol) of diethylphosphite dissolved in 1 liter of DMF is added portionwise within one hour 12.6 g (0.508 mol) of NaH (97%). The temperature is kept between 25° and 35°C. After 2 h stirring a solution of 100 g (0.254 mol) of 1benzhydryl-3-methanesulfonatoazetidine (1)⁵ dissolved in 200 ml of DMF is dropwise added. The yellow solution is stirred for 26 h at 90°C, cooled to 20°C and filtered. The solvent DMF is distilled off at 20°C/1 mbar, the residue dissolved in 1 liter of ether and three times extracted with 200 ml of H₂O each. The aqueous extracts are combined and extracted twice with 150 ml of ether each. The combined etherphases are dried with Na₂SO₄ and then the ether removed on a rotor-evaporator. The residue (69.1 g of a brown oil) is chromatographed on silica gel and eluded with ethylacetate/hexane = 3:1. There is obtained 28.9 g (31.6%) of 2a, colorless crystals, m.p. 82-84°C.

 $C_{20}H_{26}NO_3P$ (359.41) C 66.84, H 7.29, N 3.90, P 8.62% calc.: found: C 66.6, H 7.3, N 3.9, P 8.4%

¹H-NMR (in CDCl₃) $\delta = 1.25$ (t, CH₃, 6H); 2.6–3.6 (m, —CH₂—CH—CH₂—, 5H); 4.05 (quin., OCH₂, 4H); 4.4 (s, N—CH, 1H); 7.0–7.6 (m, aryl, 10H) [ppm].

2. 1-Benzhydryl-3-(diethoxymethyl-O-ethylphosphinyl)azetidine, 2b. As described for 2a, 139 g (0.708 mol) of O-ethyl-diethyoxymethylphosphonite" are treated with 17.5 g (0.708 mol) of NaH (97%) and 150 g (0.472 mol) of 1. The crude product is chromatographed on silica gel and eluded with ethylacetate/ hexane = 1:1. Yield 85.5 g (40.1%) of **2b**, yellow oil $n^{20} = 1.5360$

C 66.17, H 7.73, N 3.36, P 7.42% $C_{23}H_{32}NO_{4}P$ (417.49) calc.: found: C 65.2, H 7.6, N 3.5, P 7.0%

¹H-NMR (in CDCl₃) $\delta = 0.95-1.45$ (m, CH₃, 9H); 2.8-3.8 (m, CH₂—CH—CH₂, C—O—CH₂, 9H); 4.15 (q, P—O—CH₂, 2H); 4.48 (s, N—CH, 1H); 4.6 (d, P—CH—O), 1H, J_{PCH} 7 Hz); 6.9–7.55 (m, aryl, 10H) [ppm].

3. 3-Diethylphosphonylazetidine, 3a. 43.5 g (0.121 mol) of 2a, dissolved in 400 ml of ethanol are treated at normal pressure with H₂ in the presence of 9 g of Pd/C (5%) at ambient temperature. After 62 h H₂ up-take ceased. The catalyst is filtered and the filtrate evaporated on a rotor-evaporator. The residue (36.5 g colorless oil) is flash-chromatographed on silica gel using CH₂Cl₂/CH₂OH = 95:5 as eluent. Yield of 3a, 15.6 g (66.9%) colorless oil.

C 43.52, H 8.35, N 7.25, P 16.03% $C_7H_{10}NO_3P$ (193.10) calc.: found: C 43.0, H 8.3, N 7.1, P 15.7%

¹H-NMR (in CDCl₃) $\delta = 1.30$ (t, CH₃, 6H); 2.95 (s, NH, 1H); 3.1-4.4 (m, CH₂—CH—CH₂, OCH₂, 9H) [ppm].

4. 3-(O-ethyl-diethoxymethylphosphinyl)azetidine, 3b. 10.8 g (0.0258 mol) of 2b are treated with H₃ in the presence of Pd/C as described for 3a. There is obtained 3.86 g (59.6%) 3b, a yellow oil, $n^{20} =$ 1.4828

 $C_{10}H_{22}NO_4P$ (241.26) C 47.8, H 8.83, N 5.58, P 12.33% calc.: found: C 47.6, H 9.1, N 5.3, P 12.0%

5. 3-(O-ethyl-diethoxymethylphosphinyl)azetidine hydrochloride, 3c. To 41.2 g (0.0986 mol) of 2b in 400 ml of ethanol is added 1 equiv. HCl, 8 g of Pd/C (5%) and H₂ introduced at normal pressure (after 30% conversion another 8 g and after 70% conversion again 8 g catalyst added). After H₂ up-take ceased, the mixture is filtered and from the filtrate alcohol distilled off. The residue is dissolved in H₂O, extracted three times with ether and the aqueous phase evaporated. The residue is dried over P_2O_5 to give 24.2 g (85.5%) **3c**, a resin. $C_{10}H_{22}NO_4P \times HCl$ (287.72) calc.:

C 41.75, H 8.06, N 4.87, Cl 12.32, P10.77% found: C 39.2, H 7.9, N 4.9, Cl 12.0, P 10.5%

¹H-NMR (in DCl 10%) $\delta = 0.7-1.3$ (m, CH₃, 9H); 3.1-4.4 (m, CH₃—CH—CH₃, OCH₃. 11H); 4.85 (d, CH(OEt)₂, 1H, J_{PCH} 7 Hz); 6.95 (s, NH × HCl, 2H) [ppm].

6. Azetidine-3-phosphonic acid, 4a. A solution of 15.6 g (0.08 mol) of 3a in 150 ml of CHCl₃ is treated

with 40.7 g (0.266 mol) of Me₃SiBr and the mixture stirred for 3 days at 20° C. Then CHCl₃ is distilled off and the residue kept at 1 mbar to remove volatile material. The residue (34.5 g yellow oil) is dissolved in isopropanol and propylene oxide added. After 12 h stirring at 20° C the precipitate is filtered and washed with ether. There is obtained 10.3 g of a white solid which still contains Br. Therefore the solid is again dissolved in 100 ml of H_2 O, 40 ml of propylene oxide are added, the mixture stirred for 15 min. then evaporated at 1 mbar and the residue suspended in isopropanol/ether = 1:1. The colorless crystals are filtered and dried. There is obtained 9.1 g (82.7%) 4a, colorless crystals, m.p. $238-240^{\circ}$ C (dec.).

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C_3H_8NO_3P \times 0.1~H_2O~(138.9) calc.: C 25.9, H 5.9, N 10.0, P22.3, H<sub>2</sub>O 1.3% found: C 26.3, H 5.8, N 10.0, P21.7, H<sub>2</sub>O 1.3%
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³¹P-NMR (in D₂O) $\delta = 18.18$ ppm, J_{PCH} 11.6 Hz, J_{PCCH} 14.0 Hz

7. Azetidine-3-phosphonous acid, **4b.** A mixture of 21.3 g (0.074 mol) of **3c** and 250 ml of 2N HCl is refluxed for 5 h. The slightly yellow solution is extracted with ether and the aqueous phase evaporated at 30° C and 1 mbar. There is obtained 14.5 g of a yellow oil which is dissolved in 120 ml of H₂O, then 120 ml of propylene oxide and 30 ml of methanol are added and the mixture stirred for 2 h. Evaporation of the mixture yields a residue which is dried over P₂O₅. The sticky mass is suspended in isopropanol and stirred overnight to give 5.9 g (65.8%) **4b**, colorless crystals, m.p. 189°C (dec.)

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C_3H_8NO_2P \times 0.18 H_2O (124.32) calc.: C 28.99, H 6.81, N 11.27, P 24.92, H<sub>2</sub>O 2.61% found: C 29.1, H 6.6, N 11.0, P 24.1, H<sub>2</sub>O 2.6%
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³¹P-NMR (in D₂O) $\delta = 24.30$ ppm; J_{PH} 525.9 Hz

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